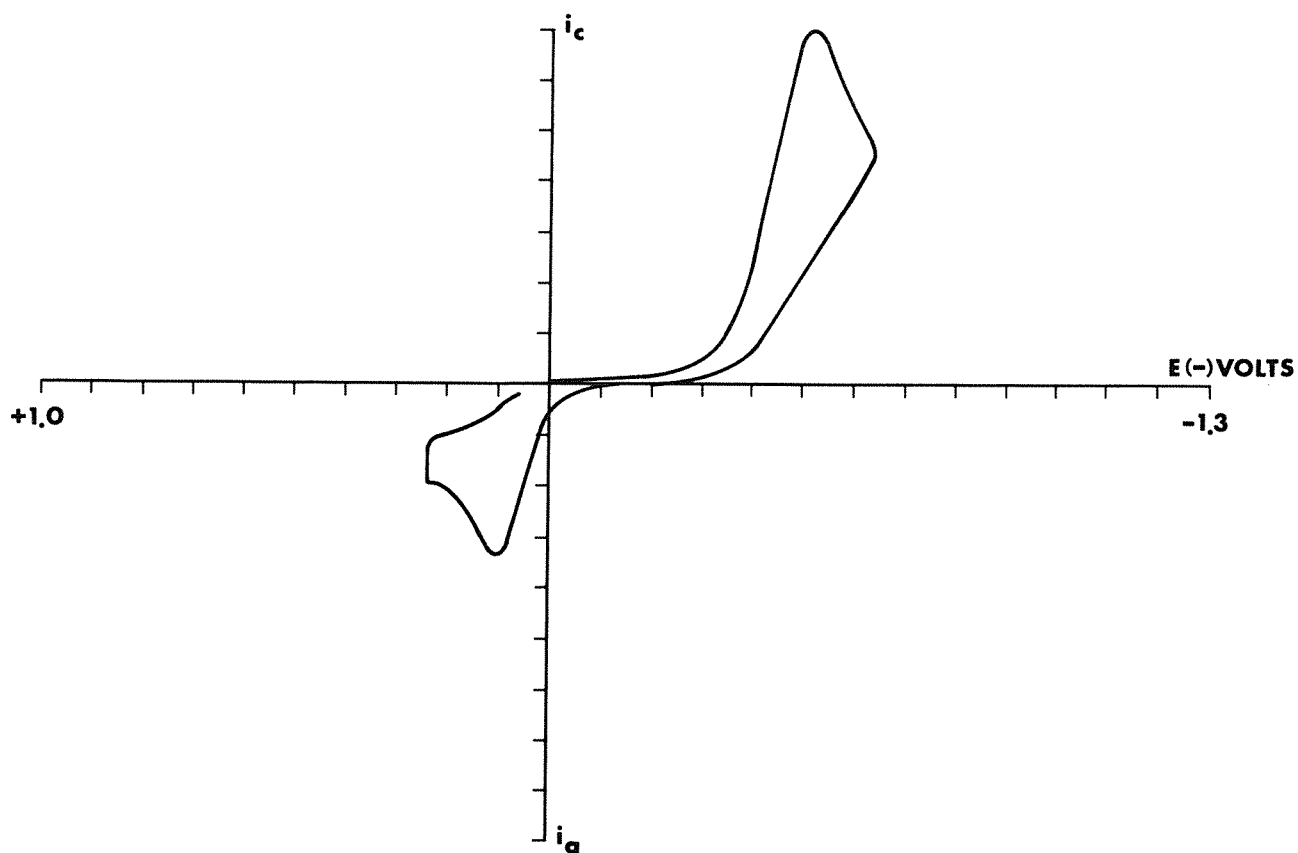
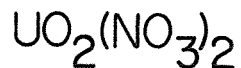


# CV NOTES

## URANYL NITRATE



SAMPLE: Uranyl Nitrate

MEDIUM: 0.05 M  $\text{HClO}_4$

CONC: 0.8 mg/ml

RATE: 200 mV/sec

ELECTRODE: Hg on Au

REFERENCE: RE-1, Ag/AgCl/DM KC1

MODEL: CV-1A

Cyclic voltammetry has proven to be very useful to the inorganic chemist. It is an invaluable tool for characterization of newly synthesized compounds. For example, if an inorganic chemist finds the reduction potential of a compound by CV, he can determine an appropriate chemical method for the same oxidation. Furthermore, he can use CV to check the purity of his new compound. This CV represents the facile reduction of +6 state of uranium. Since water is present, the reduction product is rendered irreversible by hydrolysis.



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